

Do additives affect the performance of lead-acid batteries?

This chapter reviews of the influence of additives to the pastes for positive and negative plates on the processes of plate manufacture and on the performance of lead-acid batteries. The performance of the lead-acid battery depends on the surface of the active materials of the two types of electrodes.

What is a positive electrode made of?

The composition of the alloy was the same as the positive grid produced by gravity casting. The counter electrode, with an approx. five times greater area compared to the working electrode, was made of pure lead (99.98% Pb, Avantor). Preparation of positive electrodes for the capacity test consisted of three main stages.

How to improve the performance of a lead-acid battery?

The performance of the lead-acid battery depends on the surface of the active materials of the two types of electrodes. In order to improve the performance parameters of the battery, formation of a continuous passivating  $\text{PbSO}_4$  layer should be avoided.

Can a protic ammonium ionic liquid be added to a lead-acid battery?

The proposed solution promotes the addition of a protic ammonium ionic liquid to the active mass of the positive electrode in the lead-acid battery. The experiments included the synthesis and characterisation of several protic ammonium-based ionic liquids, which differed in terms of the length of the side chain in the cation.

Can a 12V lead-acid battery be modified?

The aim of the presented study was to develop a feasible and technologically viable modification of a 12V lead-acid battery, which improves its energy density, capacity and lifetime. The proposed solution promotes the addition of a protic ammonium ionic liquid to the active mass of the positive electrode in the lead-acid battery.

How to modify lead-acid battery electrolyte and active mass?

The lead-acid battery electrolyte and active mass of the positive electrode were modified by addition of four ammonium-based ionic liquids. In the first part of the experiment, parameters such as corrosion potential and current, polarization resistance, electrolyte conductivity, and stability were studied.

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Lead-acid batteries with the positive electrode modified by  $\text{HC16SO}_4$  exhibited lower ohmic resistance than the reference. In contrast to the 2 V cells, the modified systems displayed lower values of capacitance

associated with a double layer. Charge transfer resistance and diffusion coefficients remain almost at the same level.

In this paper, the positive additives are divided into conductive additive, porous additive and nucleating additive from two aspects: the chemical properties of the additives and ...

The positive active-material of lead-acid batteries is lead dioxide. During discharge, part of the material is reduced to lead sulfate; the reaction is reversed on charging. There are three types of positive electrodes: Plant&#233;, tubular and flat plates. The Plant&#233; design was used in the early days of lead-acid batteries and is still produced today for certain ...

The positive electrode is one of the key and necessary components in a lead-acid battery. The electrochemical reactions (charge and discharge) at the positive electrode are the conversion ...

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Electrochemical study of lead-acid cells with positive electrode modified with different amounts of protic IL in comparison to unmodified one, (a) discharge curves of selected cells at current ...

Enhancement of the discharge capacity and cycle life of lead-acid batteries demands the innovative formulation of positive and negative electrode pastes that can be ...

Although descriptions of Pb-acid cells always say that the negative electrodes are primarily lead, and the positive electrodes primarily  $PbO_2$ , they are both initially made from the same material, a paste consisting of a mixture of  $PbO$  and  $Pb_3O_4$ . It can be considered to be lead powder that is 70-85% oxidized, and is traditionally called "leady oxide". Measured ...

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The positive electrode is one of the key and necessary components in a lead-acid battery. The electrochemical reactions (charge and discharge) at the positive electrode are the conversion between  $PbO_2$  and  $PbSO_4$  by a two-electron transfer process. To facilitate this conversion and achieve high performance, certain technical requirements have to ...

The structure and properties of the positive active material  $PbO_2$  are key factors affecting the performance of lead-acid batteries. To improve the cycle life and specific capacity of lead-acid batteries, a chitosan (CS)-modified  $PbO_2$ -CS-F cathode material is prepared by electrodeposition in a lead methanesulfonate system.

Dietz H, Garche J, Wiesener K (1987) On the behaviour of carbon black in positive lead-acid battery electrodes. *J Appl Electrochem* 17(3):473-479. Article CAS Google Scholar Ball RJ, Evans R, Thacker EL, Stevens R (2003) Effect of valve regulated lead/acid battery positive paste carbon fibre additive. *J Mater Sci* 38:3013-3017

In this paper, the positive additives are divided into conductive additive, porous additive and nucleating additive from two aspects: the chemical properties of the additives and the effect on the performance of the lead-acid battery. The effect and mechanism of different additives on the structure and properties of positive electrode are ...

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The addition of single-wall carbon nanotubes (SWCNT) to lead-acid battery electrodes is the most efficient suppresser of uncontrolled sulfation processes. Due to the cost of SWCNT, we studied...

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